1) Modelling an inversion

Material
- salted water (300g of salt per liter) at 5°C
- hot fresh water at 50°C
- tall beaker
- plastic film (useful for pouring hot water gently on top of cold salted water without mixing it; you get rid of the film gently after that)
- colour Tepid fresh water filled in small bottle at room temperature.
- a metal wire attached to the flask: useful for pushing down the small bottle at the bottom of the beaker.

I) Pour cold, salted water in the beaker up to mid-height

II) Cover the water with the plastic sheet

III) Pour gently hot fresh water on the top of the filmed layer of salted water, in order to avoid mixing the two water masses.

IV) Pull gently the film, avoiding turbulence.

V) Put the small bottle filled with colored tepid water at the bottom of the beaker (using the wire); incline the opening of the bottle so as to let the colored water go out gently.

Questions

1.1) Indicate which drawing best describes what you observe in your beaker (5 points)
Ans. ( )

A | B | C | D
1.2) Please indicate the relative densities of the 3 water masses
(HF : hot and fresh, CS : cold and salted, TF : Colour Tepid fresh) : ( 2 points)
Ans. (         )

A) density\_CS > density\_HF > density\_TF
B) density\_HF > density\_CS > density\_TF
C) density\_TF > density\_CS > density\_HF
D) density\_CS > density\_TF > density\_HF

2) Inversion in a valley

Inversion can be a problem when there is pollution. Let’s consider a valley with a polluting factory in winter.

2.1) Associate each time of the day/weather condition below with one of the vertical profiles of temperature A, B, and C in the small box ( 3 points)

2.2) Draw the limit of the polluted layer on the valley profile above (if any) ( 2 points)
An emagram (Skew T Log-P-diagram in the US) is a thermodynamic diagram which allows to predict how the state of a parcel of air changes when the parcel moves up or down. You may kindly note that in those diagrams, isotherms (solid brown lines) are skewed towards the right at 45°, so you must read temperature obliquely! Pressure/altitude is on the vertical axis.

We are going to apply it to the study of a famous Argentinian wind, blowing from the Andes: Zonda wind.

Let’s consider a parcel of humid air arriving from the Pacific Ocean in Chile. At the altitude of 200 meters, air pressure is 1000 hPa. The temperature of this parcel of air is then 15°C (A point) and its mixing ratio (water content) is 6 g of water per kg of air.
3.1) Plot this point (A point) on the emagram. Plot the dew point (D_A) at this altitude and indicate the dew point temperature. (1 point)

Ans.

West winds push this parcel of air up the Chilean slopes of the Andes. We will consider that this move is adiabatic: the parcel of air does not exchange heat with the surrounding air and cools just because its pressure decrease. At some point, the air parcel will reach saturation and rain will start (B point).

3.2) Plot the point B up to which the rain starts (saturation) on the emagram and draw the path between A and B. (1 point)

Now saturated air continue to move up the slope, until the summit of an Andean mountain pass, at 2500 m (750 hPa), (C point)

3.3) Plot the point C giving the state of the parcel of air at 750 hPa, and draw the path between B and C. (2 points)

Now the parcel will go down the argentinian slope of the Andes, until it arrives at the altitude of 200 m (1000 hPa), (D point)

3.4) Plot the point D giving the state of the parcel of air at 1000 hPa, and draw the path between C and D. Indicate the temperature and mixing ratio of the air in D. (2 points)

Ans.

3.5) Choose how to describe the perception of the Zonda wind in Argentina (1 point)

(A) warm and humid
(B) cold and humid
(C) warm and dry
(D) cold and dry.

Ans.
Emagram (Skew-T Log P)

Moist adiabatic lapse rate: shows the cooling of a parcel of moist air when uplifted

Dry adiabatic lapse rate: shows the cooling of a parcel of dry air when uplifted